Ethyl acetate ext Pedaliin (m p and mixed m p, acid hydrolysis to yield pedalitin and glucose, R_f and co-chromatography with authentic samples).

Aqueous mother-liquor Added equal volume of 14% H_2SO_4 , kept in a boiling water bath for 2 hr, cooled and extracted with ether and EtOAc EtOAc concentrate on dilution with acetone yielded a yellow solid, mp 185° , found to be a mixture of two glycosides (PC, R_f 0.45 and 0.60, H_2O) The pigments were separated by column chromatography on cellulose.

Dinatin-7-glucuronide. (R_f 0.45, eluted with water-saturated EtOAc containing 5-25% MeOH), decomposing at 221-223°, λ_{max} MeOH 270, 337 nm (no shift of Band II with NaOAc), hydrolysis with 10% H₂SO₄ in HOAc medium for 5 hr yielded dinatin and glucuronic acid

Diosmetin 7 glucuronide. (R_f 0.60, eluted with MeOH), decomposing at 210–212°, λ_{max} 268, 340 nm (no shift of Band II with NaOAc), hydrolysis as above yielded diosmetin and glucuronic acid

Comment. This is the first record of isolation of a dinatin glycoside and diosmetin glucuronide. Considerable difficulty was experienced in separating the glucuronides of the two isomers by fractional crystallization. The occurrence of dinatin and pedalitin in the Pedaliacea is significant from the point of view of chemotaxonomy in view of the frequent records of 6-hydroxy or 6-methoxyflavones in families of the Tubiflorae ^{2,3} Dinatin glucuronide was also detected in the leaves of Sesamum indicum (Pedaliaceae) in addition to pedalitin and pedaliin recorded earlier.²

³ J B HARBORNE and C A WILLIAMS, Phytochem 10, 367 (1971)

Key Word Index-Pedalium murex, Pedaliaceae, flavonoids, pedalitin, dinatin; diosmetin.

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SAPOTACEAE

CONSTITUENTS OF MADHUCA LONGIFOLIA LEAVES

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Plant. Madhuca longifolia syn. Bassia longifolia. Uses Medicinal, seed fat a commercial commodity, flowers rich in sugar—foi fermentation (country liquor) 3,4 For all practical purposes identified with Madhuca indica. Previous work. Madhuca latifolia. Seed fat, seed saponin, 7,8 blossoms, 3,4 bark 9

- ¹ B Mukerji, Indian Pharmaceutical Codex, Vol I, p 144, CSIR (India), New Delhi (1953)
- ² Wealth of India, Vol VI, p 210, CSIR (India), New Delhi (1962)
- ³ G J FOWLER, G D E BERHAM, S N BHATE, K H HASSAN, H MEHDIHASSAN and N N INUGANTI, J Indian Inst Sci 3, 81 (1920)
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- ⁵ P Van Rayan, Blumea, 10, 122 (1960)
- ⁶ T P HILDITCH and M B ICCHAPORIA, J Chem Soc Indian 57, 44T (1938)
- ⁷ B J HEYWOOD, G A R KON and L WARE, J Chem Soc 1124 (1939)
- ⁸ B J Heywood and G A R Kon, J Chem Soc 713 (1940)
- ⁹ Y C. Awasthi and C R Mitra, Phytochem 7, 1432 (1968)

Leaves. EtOH extr. fractionated successively with n-hexane, ether and ethyl acetate.

Hexane extract Chromatography, neutral Al_2O_3 β -D-glucoside of β -sutosterol, m p, mixed m p, acetate, m.p, mixed m p, and superposable IR, hydrolysed to β -sutosterol and glucose with 5% H_2SO_4 (alc) Stigmasterol, m.p., mixed m.p, acetate, m p, mixed m p, superposable IR and co-TLC n-Hexacosanol, m p, mixed m p, acetate, m p, mixed m p and IR.

New triterpene esters. Chromatographic fractions (benzene-hexane, 1 4) yielded an L-B and Noller's positive compound, m p. 152°, C₄₀H₆₈O₃ (Found C, 80·22, H, 11·52 Required C, 80.53, H, 11.40%), ν_{max} 3448 (free OH), 1709 (ester carbonyl) and 820 (trisubstituted double bond) cm⁻¹ The ester on hydrolysis with 5% alcoholic KOH yielded erythrodiol, C30H50O2, mp, mixed m.p, acetate, mp, mixed mp, superposable IR and co-TLC, and a fatty acid identified as capric acid, C₁₀H₂₀O₂, mol wt 172 (M⁺), neutr equiv 170 Mass and NMR data of the ester favoured the attachment of the capric acid moiety at 3β —OH of erythrodiol and subsequently the structure of the new triterpene ester was assigned as 3β -caproxy-olea-12-en-28-ol In the mass spectra, the molecular ion peak was obtained at m/e 596 and a loss of 172 mass units ($C_{10}H_{20}O_2$) was recorded in the appearance of a peak at m/e 424, which on subsequent loss of 31 mass units (-CH₂OH) gave a prominent peak at m/e 393 The characteristic retro-Diels-Alder fragmentation pattern was observed with a peak at m/e 234 arising due to 'right half' of Δ^{12} -oleananes ¹⁰ Loss of 31 mass units (m/e 203, base peak) was also observed from this fragment which further substantiated the presence of free -CH₂OH at C-28 Other prominent peaks were at m/e 189, due to 'left half' of Δ^{12} -oleananes¹¹ and at m/e 133, due to further fragmentation of m/e 203 fragment in ring-E 10

The NMR spectra of the ester further confirmed the presence of free —CH₂OH at C-28 as an AB type quartet (2H) centred around τ 6 65 (J=11 c/s) appeared due to methylene protons of free primary hydroxyl group,¹² presence of α -proton at C-3 (1H) of 3β -acyloxy oleananes was observed as a triplet centred at τ 5 4 (J=8 c/s)¹² and olefinic proton of Δ^{12} -oleananes was observed as an unresolved multiplet centred at τ 4 7 (1H)

Oleanolic acid palmitate The chromatographic fractions (CHCl₃-MeOH, 3 1) yielded another L-B and Noller's positive ester, m.p 145°, $C_{46}H_{78}O_4$, mol. wt 694 (M⁺), ν_{max} 1739, 1709 (ester carbonyl and carboxyl) showed prominent peaks in the mass spectra at m/e 438 (M⁺ - 256), m/e 248 (right half retro-Diels-Alder fragment), m/e 203 (248-COOH) and NMR peaks at τ 47 and τ 5 45 (see erythrodiol caprate vide supra) On hydrolysis with 5% alcoholic KOH, the ester gave palmitic acid, m p, mol wt 256 (M⁺) and oleanolic acid, m p, mixed m p of acid and its methyl ester, IR and co-TLC. Oleanolic acid palmitate in crude form has earlier been reported from Madhuca butyracea. 14

Ether fraction Traces of xanthophyll, m p 186°, $\lambda_{\text{max}}^{\text{cal}}$ 424, 446, 472 nm, IR 3390 cm⁻¹ (free OH). The mass spectra of the compound could not be recorded due to its decomposition Flavonoid m p > 325°, $C_{15}H_{10}O_8$, mol wt 318 (M⁺), $\lambda_{\text{max}}^{\text{cal}}$ 254, 378, with anhydrous aluminium chloride, $\lambda_{\text{max}}^{\text{alc}}$ 266, 435 nm, penta-acetate, (M⁺ = 528), m p 191°, hexamethyl ether, (M⁺ = 404), m.p 155° Though colour reactions (Shinoda¹⁵ and alkali) and UV

¹⁰ C R Noller, R A Smith, C H Harris and J W Walker, J Am Chem Soc 64, 3047 (1942)

¹¹ H BUDZIKEWICH, J M WILSON and C DJERASSI, J Am Chem Soc 85, 3688 (1963)

¹² J B THOMPSON, Tetrahedron 22, 351 (1966)

¹³ G N PANDEY and C R MITRA, Tetrahedron Letters 15, 1352 (1967)

¹⁴ Y C Awasthi and C R Mitra, Phytochem 7, 637 (1968)

¹⁵ J Shinoda, J Pharm Soc Japan 48, 214 (1928)

data indicate its similarity with myricetin, ¹⁶ absence of any shift in the Band II of the UV spectrum on addition of anhydrous sodium acetate indicates absence of free hydroxyl at C-7.

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Key Word Index—Madhuca longifolia, Sapotaceae, erythrodiol caproate, oleanolic palmitate, triter-penoid esters, Sapotaceae.

¹⁶ T A GEISSMAN, in Modern Methods of Plant Analysis (edited by K PEACH and M V TRACEY), Vol III, p 464, Springer, Berlin (1955)

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SOLANACEAE

BOMBIPRENONE FROM NICOTIANA TABACUM

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Abstract—Bombiprenone (6,10,14,18,22,26,30,34-octamethyl-5,9,13,17,21,25,29,33-pentatriacontaoctaen-2-one) has been isolated from flue-cured tobacco (*Nicotiana tabacum*) in yields of 0.1-0.6% dry wt. Plasto-quinone-A was converted to bombiprenone by aerobic photo-oxidation, most probably via a hydroperoxide

A LIPID component isolated from flue-cured tobacco by a procedure involving hexane extraction, column chromatography on Florisil, and preparative-layer chromatography on silica gel G has been identified as 6,10,14,18,22,26,30,34-octamethyl-5,9,13,17,21,25,29,33-pentatriacontaoctaen-2-one (I)

In 1960, Kofler et al.¹ prepared (I) as an intermediate for the total synthesis of solanesol Eight years later, (I) was isolated in 0.4% yield from the unsaponifiable matter obtained from the lipid extract of the faeces of the silkworm (Bombyx mori, L) by Toyoda et al.² who proposed the trivial name bombiprenone Bombiprenone was also isolated in 0.002% yield from the leaves of mulberry (Morus bombycis, K.), which is the sole diet of the silkworm

¹ R RUEGG, U GLOOR, A LANGEMANN, M KOFLER, C VON PLANTA, G RYSER and O ISLER, Helv. Chim Acta 43, 1745 (1960)

² M TOYODA, H FUKAWA and T SHIMIZU, Tetrahedron Letters 3837 (1968)